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by H.B. Santoso

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Research Article

Iron bioaccumulation and ecological implications in the coastal swamp wetlands ecosystem of South Kalimantan: Insights from giant mudskipper fish as bioindicators

Heri Budi Santoso^{1*}, Krisdianto¹, Rizmi Yunita²

- Department of Biology, Faculty of Mathematics and Natural Sciences, Lambung Mangkurat University, Banjarbaru, South Kalimantan, Indonesia
- ² Department of Aquatic Resources Management, Faculty of Fisheries and Marine Science, Lambung Mangkurat University, Baniarbaru, South Kalimantan, Indonesia

*corresponding author: heribudisantoso@ulm.ac.id

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Abstract

This study investigated iron (Fe) accumulation in South Kalimantan's coastal swamp wetlands ecosystem, utilizing giant mudskipper fish as indicators of heavy metal pollution. By analyzing Fe levels in water, sediment, and fish organs over time, insights into its effects on the environment and human health were gained. Furthermore, through Atomic Absorption Spectrometry, Fe concentrations in Kuala Lupak's coastal wetland were measured. revealing significant correlations between Fe levels in water, sediment, and fish tissues. These results enhance our understanding and inform better management strategies. Anthropogenic and natural sources contribute to the accumulation of heavy metals, particularly Fe, with anthropogenic pollution being the most dominant. This study presented the escalating concentrations of Fe within the Kuala Lupak estuary and raised concerns regarding the ecological and human health implications. Continuous monitoring, source identification, public awareness, regulations, remediation, and long-term exploration are essential for addressing heavy metal pollution and its ecological impact. Therefore, valuable insights are provided for environmental management and conservation efforts.

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Introduction

Heavy metals are a significant threat to the health of estuarine ecosystems, affecting the biodiversity of aquatic organ 35 is at both individual and community levels (Sarah et al., 2019; Tabrez et al., 2021; Zaoui et 18 2023). This negative impact is attributed to the bioaccumulation of heavy metals in aquatic biota, resulting from the discharge of industrial and domes 27 waste into the estuarine environment. Yogeshwaran et al. (2020) reported that the bioaccumulation of the metals in estuarine a 39 its biota is directly linked to water contamination. The presence of heavy metals in aquatic biota is due to their availability in water bodies

and sediment, as well as the food consumed by the fish (Moniruzzaman and Saha, 2020). These trace elements exist in sediments in various forms, including dissolved, bound to suspended particulate matter, or accumulated, enhancing their incorporation into biological pre 50 ses (Soltani et al., 2021).

Iron, as the fourth most 4 minant metal on Earth, possesses the capacity to change valence and form complexes with oxyger 61 rthermore, it plays a crucial role in respiration in nearly all aerobic organisms. 14 wever, unless suitably sequestered, Fe can catalyze the formation of radicals capable of damaging biological molecules, cells, tissues, and entire organisms (Tenenbein and Huang, 2022). Certain trace

38 tals such as Fe, Cu, Mn, and Zn are essential to biological systems but can become to at high concentrations. Metals, including Zn, Fe, Cu, and Mo, are integral components of metal-protein complexes in enzymes, contributing to enzymatic activities. Fe is a constituent of oxygen transport within living organisms, while it also plays a vital role in oxidative metabolism, serving as a cofactor for proteins and enzymes, DNA synthesis, and electron transport (Zafar and Khan, 2020).

Other heavy metals such as Cd, Cr, Pb, and Hg have adverse effects even at deficient concentrations (Akinsanya et al., 2020). Despite its physiological functions in fish, environmental pollution can render Fe toxic due to excessive accumulation (Garg and Gauns, 2023). The continuous and substantial discharge of trace metal pollutants into the sea poses a phificant threat to estuarine and coastal ecosystems due to their chronic toxicity, non-biodegradability, and bioaccumulation in biota (Zaynab et al., 2022). The metals can be transferred and biomagnified through food chains, presenting significant risks to human health. Metal contamination in estuarine ecosystems can induce toxic effects in fish and humans, primari 26 by consuming contaminated food (Franco-Fuentes et al., 2023; Prasad et al., 2023). Therefore, evaluating the occurrence of Fe content in the estuarine ecosystem of South Kalimanian is essential.

Analyzing the levels of Fe in the water body, sediments, and its bioaccumulation in wild fish, particularly those that inhabit estuaries for a significant portion of their life, such as the giant mudskipper fish (*Periophthalmodon schlosseri*), is 12 great importance. This fish species holds potential as a bioindicator for monitoring heavy metal pollution in estuarine waters, aiming to mitigate the adverse impacts of Fe on biodiversity conservation and environmental health. Evaluating the presence of Fe in the ecosystem serves as a valuable environmental indicator to understand anthropogenic influences and associated risks to the estuary environment.

Iron is an abundant transition metal in Earth's crust (Jaishankar et al., 2014), and it can exist in aquatic environments in dissolved or suspended forms in water or sediment. Investigating Fe in water, sediment, and fish tissues aids in determining contamination status and loads. Therefore, quantifying the presence of this metal is pivotal in analyzing the health of aquatic ecosystems. Valuable information about the impacts of environmental changes and anthropogenic activities over time is obtained through long-term bioaccumulation of heavy metals. As a result, biomonitoring studies are essential in tracking trends in the concentrations of these metals, enabling scientists to identify changes in estuarine health (Muhammad and Ahmad, 2020). Contamination of aquatic ecosystems by trace metals has garnered global attention from environmental and biological experts, leading to numerous studies on their accumulation in estuarine and marine biota (La Colla et al., 2017).

There is a pressing need for comprehensive information on the evaluation of Fe presence, toxicity, and effects on giant mudskipper fish in the South Kalimantan estuary. Therefore, effective monitoring and evaluation of the availability of this metal in the estuarine environment (water, sediment, biota) are crucial. The giant mudskipp 16 ish has potential as a bioindicator species for heavy metal pollution biomonitoring programs in the estuarine ecosystem of South Kalimantan because it meets the requirements, including natural abundance (Hidayaturrahmah et al., 22 9), easy to obtain in large quantities, wide tolerance to polluted environmental conditions, benthic living habits, high trophic level in the aquatic food chain, and capability to accumulate heavy metals (Santoso et al., 2020). Fish, renowned for their capacity to accumulate heavy metals, are well-suited to serve as bioindicator species in pollution biomonitoring, enabling the assessment of estuarine ecological health (Alizada et al., 2020). The giant mudskipper fish, in particular, possesses unique biological characteristics, namely having amphibian-like characters, which distinguish them from other species. Additionally, they are often called "fish walking on land" or "fishes out of water"(Kumaraguru et al., 2020).

Kuala Lupak in the Barito River estuary is a coastal swamp wetland area characterized by extensive transportation activities, loading and unloading of coal, and coal mining waste disposal. The continuous input of waste from the coal mining process and coalrelated activities into the water body has the potential to cause estuary pollution when prolonged and accumulated. These waste materials contain heavy metals such as Mn, Fe, Hg, Cd, Cr, Pb, and Cu (Kasmiarti et al., 2021). Furthermore, Kuala Lupak has witnessed the conversion of mangrove land to ponds, a place for ship transportation traffic and settlements. Several industries along the riverbank, such as wood/plywood processing factories, rubber, fisheries, and oil palm plantations, pollute estuary waters through waste discharge (Sofarini et al., 2012). It is hypothesized that the unchecked release of these wastes into water bodies poses a significant risk to both aquatic ecosystems and human health. Therefore, urgent biomonitoring efforts are warranted for early detection of the negative imp 48 of pollution. In light of these considerations, it is important to investigate the presence and occurrence of Fe in the Kuala Lupak estuary ecosystem during the period from 2020 to 2022.11

The objectives of this study were to evaluate the presence of Fe levels in water 57 dies, sediments, and giant mudskipper fish, assess the bioaccumulation of Fe in various tissues of giant mudskipper fish (liver, kidney, muscle, skin, and gills), evaluate the relationship or correlation between Fe content in water leaves and sediments as well as the concentrations in the liver, kidney, skin, muscle, and gill tissues of giant mudskipper fish, and validate the potential of giant mudskipper fish as a bioindicator of Fe contamination

in the Kuala Lupak estuary. The trends and intricacies of iron (Fe) contamination are investigated into by this research, especially within estuarine environments. The accessibility of this metal was primarily evaluated, and its annual fluctuations were also monitored. The knowledge acquired from this study provides significant insights, especially for comprehending the possible consequences of future environmental changes. The results of this multi-year trace metal monitoring are of paramount importance in understanding the bioaccumulation potential of trace metals and constitute a crucial step in the management of estuarine and coastal ecosystems. These outcomes serve as a foundational framework for stackholders to assess and address anthropogenic effects in the future.

Materials and Methods

Study area

The study was conducted in the coastal wetland of Kuala Lupak, situated in the Barito Kuala Regency. This region is approximately 8 km west of the mouth of the Barito River, the largest and longest river in South Kalimantan Province, Indonesia. The coastal wetland area of Kuala Lupak directly merges with the Java Sea coastline to the south. A comprehensive review of each study station/site is presented in Table 1, while the precise sampling locations are visually shown in Figure 1.

Sample collections and analysis

The process of collecting water, sediment, and giant mudskipper fish samples was conducted during August in 2020, 2021, and 2022. The sampling was carried out using the purposive sampling method, where sample selection was based on specific criteria, such as the potential sources of pollution. To accurately identify these sources, various activities that contribute to contamination were taken into consideration. Figure 1 shows the collection of fish, water, and sediment samples at the study location.

Water

Using an Aqua trap water sampler, surface water samples were collected from two stations scattered throughout the study reg 3n. In preparation for further analysis, approximately 2 mL of concentrated HNO₃ was added to ead L sample after filtration. The study adopted the liquid-liquid extraction technique (Jayaprakash et al., 2015) to determine the ncentration level of dissolved Fe. A 100 mL unfiltered sample was placed in a separating funnel and mixed with 2 mL of 2% ammoniu 40 yrrolidine dithiocarbamate (APDC). Subsequently, 10 mL of isotyl methyl ketone (IBMK) was used for extraction after severe agitation. The extracted aqueous phase 3 as subjected to further extraction using concentrated HNO₃ and high-purity water. Any remaining traces of organic solvents in the solutes were evaporated on a low-temperature hot plate, and the final solution was analyzed using Atomic Absorption Spectrometry (AAS Thermo Tentific ICE 3500 series Germany). This instrument was used to measure the concentration level of Fe in water, sediments, and fish samples. For water, the concentration was expressed in mg L⁻¹. Finally, samples for metal determination were conditioned following Jayaprakash et al. (2015).

Sediment

Sediment samp were procured from two estuary stations using a Van Veen grab sampler. Furthermore, they were dried at 40°C and powdered in an agate mortar for the digestion procedures. Approximately 1 g of the dried sediment sample was digested with an acid mixture containing HClO₄+HF+HCl, and the residue was dissolved 1th concentrated HCl, then diluted to 25 mL. The solution was analyzed using flame AAS to estimate the Fe concentrations which were expressed in mg kg⁻¹ for sediment.

Fish sampling and analysis

Fish samples were obtained from the estuarine area by local fishermen. A total of 15 giant mudskipper fish with identical dimensions, ranging between 19 and 20 cm in length and weighing from 160 to 170 g, were collected from each sampling statum to ensure representativeness. The obtained samples were transported to the laboratory in ice boxes to maintain their preservation and appr18 riate conditions for further analysis. Furthermore, the dissection of the fish was conducted using surgical stainless-steel instruments, facilitating the extraction of organs, including the liver, kidneys, gills, muscles, and skin. A digestion tube, pre-loaded with 5 mL of HNO3 and H2SO4, was utilized for each organ sample to initiate f4 reaction. The samples were subjected to digestion using a hot block digestion apparatus, maintained at 60°C for 3(53 inutes. Following the cooling of the samples, 10 mL of HNO3 was added, and the mixture was further heated at temperatures ranging from 120 to 150 °C until a discernibly dark coloration of the solution was achieved. For facilitating the subsequent filtration process, 1 mL of 12O2 was introduced to yield a clear solution. Those filtered samples were then analyzed using Atomic Absorption Spectrometry (AAS) to quantify the concentration of Fe in the fish tissues, expressed mg kg-1.

Statistical analysis

Data processing and analysis were executed employing GraphPad Prism software, specifically the 5.9.0 series from Dotmatics. The statistical evaluations comprised the determination of mean, standard deviations (SD), minimum and maximum values, the creation of graphical representations showing the interrelationship among variables, and the calculation of Pearson correlations. The analyses were conducted using GraphPad Prism 7.02, developed by the company of GraphPad Software Inc., San Diego.

Table 1. Study station description.

Location	Stations/sites	Description		
Kuala Lupak	Estuarine	The estuarine water of Kuala Lupak is a coastal wetland area with the status of Margasatwa Wildlife Sanctuary, specifically a mangrove swamp ecosystem. The mouth of the Kuala Lupak River is surrounded by human settlements, and it flows into the Java Sea. The samples were collected from the coordinates 3°27'28.223"S 114°22'04.780" E in the estuarine waters.		
	Coastal	The coastal waters of Kuala Lupak are located approximately 8 km west of the mouth of the Barito River and border the Java Sea to the south, with a coastline of 30 km. Furthermore, they are characterized by mangrove swamps. At the coordinates 3°28′02.383″S 114°21′31.282″ E samples were collected in the coastal waters where the Kuala Lupak River flows into the sea.		

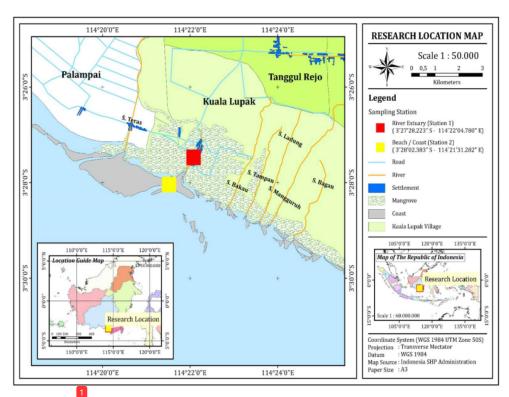


Figure 1. Study area map and sampling locations in the estuary Kuala Lupak South Kalimantan.

Results Occurrence of Fe concentrations in water and codimont

The average concentrations of Fe in the water column from 2020 to 2022 ranged from 2.55 to 11.61 mg L⁻¹. This showed a significant trend of increase over each successive year. The highest average concentration of Fe was recorded in 2022 to 37: 11.61 mg L⁻¹, as presented in Table 2. The Maximum Permissible Limits (MPLs) for heavy metals in estuary water were

not regulated by International Guideline values. The Fe in Kuala Lupak estuary water exceeded the fe in Kuala Lupak estuary water exceeded the following the set by Government Regulation of Indonesia number 22 of 2021 concerning the Implementation of Environmental Protection and Management. Furthermore, the seawath 19 quality standard for marine biota was set at 0.30 mg L⁻¹. In parallel, the average concentrations of Fe within the sediment, serving as 49 e habitat of the giant mudskipper fish, was in the range of 1,245.92 to 48,911.78 mg kg⁻¹.

Analogo to the pattern observed in the water column, the concentrations of Fe in the sediment also displayed a significant upward trend over each year, as shown in Figure 2. The highest average level of Fe (48,91 63 mg kg⁻¹) in the sediment was observed in 2022 (Table 2). The concentration of this metal 52 Kuala Lupak estuary sediments exceeded the maximum limit of 20,000 mg kg⁻¹ set by the

24 ernational Guideline value, CBSQG 2003: Consensus-Based Sediment Quality Guidelines (Wisconsin Department of Natural Resources). The mean plot of Fe level in water and sediment from 2020 to 2022 demonstrated a significant interrelationship between variables, with a confidence level of 95%, as indicated by the statistical analysis $(p<0.05, R^2=0.649)$.

Table 2. Mean values of Fe concentration in water, sediment, and some tissues of giant mudskipper fish from the Kuala Lupak estuary between the period of 2020 to 2022.

Sampling Period	13 ter (mg L ⁻¹)	Sediment (mg kg ⁻¹)	Liver (mg kg ⁻¹)	Kidney (mg kg ⁻¹)	Muscle (mg kg ⁻¹)	Skin (mg kg ⁻¹)	Gill (mg kg ⁻¹)
2020	2.55	1,245.92	5.52	7.98	40.44	4.42	1.23
2021	5.54	47,348.42	8.43	11.99	51.56	6.70	3.32
2022	11.61	48,911.78	12.84	15.44	69.74	8.80	6.22

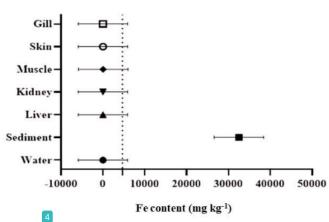


Figure 2. The mean plot of Fe concentration in water, sediment, and some tissue of giant mudskipper fish from 2020 to 2022.

Temporal variations in Fe concentrations in giant mudskipper fish

The mean Fe concentrations in the liver, kidney, muscle, skin, and gills of giant m 10 kipper for the years 2020-2022 ranged from 5.52 to 12.84 mg 10 11, 7.98 to 15.44 mg kg⁻¹, 40.44 to 69.74 mg kg⁻¹, 4.42 to 8.80 mg kg⁻¹, and 1.23 to 6.22 mg kg⁻¹, as presented in Table 2. The average level of this metal in several fish tissues exhibited a significant upward trend annually, with the highest value being recorded in 2022, as specified in Figures 2 and 3. Based on the results in Table 2, muscle had the highest accumulation among the tested fish tissues, followed by kidney, liver, skin, and gills. The Maximum Permissible Limit (MPL) for Fe in fish meat was not regulated by international guideline values but was regulated by national guideline values. The Fe in giant mudskipper fish organs exceeds 12 threshold value of 1 mg kg-1, as regulated by the Indonesian National Standard (SNI 7387:2009) regarding the maximum limit of heavy metal contamination. When comparing the content of this metal among the tested samples in the Kuala Lupak estuarine ecosystem, the average concentration in sediment was the highest.

The relationship between Fe contents in water, sediment, and giant mudskipper fish tissues

among interrelationships heavy concentrations in water, sediment, and the tissues of giant mudskipper fish from 2020 to 2022 were evaluated and presented in Figur 17. The graphical representation shows a substantial positive correlation between the concentration of Fe in the water and the levels identified in the liver, muscle, and gills, with a correlation coefficient of 1. This implied that elevated levels of this metal in the water correspond with its heightened concentrations in the liver, muscles, and gills. Furthermore, a robust positive correlation was observed between waterborne Fe concentrations and the levels of the metal in the kidney and skin, with correlation coefficients of 0.97 and 0.98, respectively.

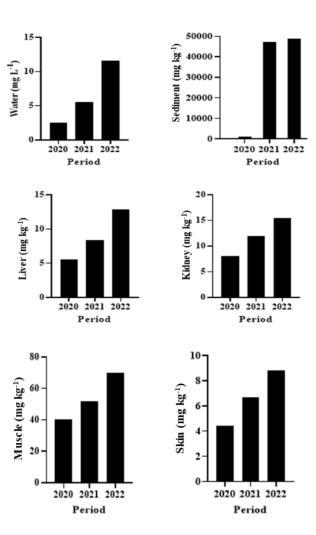


Figure 3. Fe content trend in some tissue of giant mudskipper fish period 2020-2022.

The correlation coefficient between Fe levis in the water and sediment was 0.77. A highly significant 9 sitive correlation among the concentrations present in the water, sediment, and tissues of the giant mudskipper fish, as indicated by the outcomes of this investigation. Based on these observations, the accumulation in water and sediment has augmented Fe levels within the tissues of the fish. This correlation was in line with the consistently high concentrations in water and sediment. Figure 3 further shows a robust positive correlation between sediment-bound Fe concentrations and those in the kidney, skin, gills, liver, and muscle tissues, with correlation coefficients of 0.90, 0.89, 0.83, 0.82, and 0.81, respectively. These results underscored the effect of sediment-bound Fe accumulation on the corresponding values within the tissues of the fish. This correlation was in line with the elevation of sediment-bound concentrations over 3 years, exhibiting an annual increment, as presented in Figure 1. The study also underscored a significant positive correlation across various organs of the giant mudskipper fish. This indicated a uniform distribution across diverse organs after Fe uptake into the body of the fish. In Figure 3, a close positive correlation is evident between the Fe concentration in the skin, serving as an entry point for Fe uptake. In the kidney, gills, muscle, and liver, the level of this metal exhibited correlation coefficients of 1.00, 0.99, 0.99, and 0.99, respectively. Similarly, a positive, strong correlation existed among the concentration in the gills, another site of Fe entry, and the Fe levels in the liver, muscle, skin, and kidney.

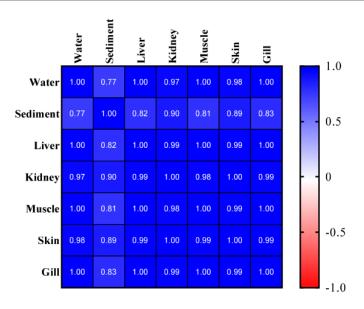


Figure 4. Pearson correlation of Fe content in some tissue of mudskipper fish period 2020-2022.

These featured co 59 cients of 1.00, 1.00, 0.99, and 0.99, respectively. The results showed that elevated Fe concentrations in the skin and gills precipitate were directly proportional to the liver, kidney, and muscle tissues.

Discussion

Fe contamination in estuarine waters: Sources and implications

The heightened prevalence of Fe within the Kuala Lupak estuary in 2022 was attributed to the effects of extreme rainfall and substantial flooding in the region of Kalimantan Selatan from December 2020 to February 2021. This event led to the dissolution of heavy metal deposits from upstream areas, resulting in their transport into the estuarine en 340nment and eventual deposition within the estuary. This result was in line with the study conducted by (Čmelík 1111., 2019), which identified a substantial increase in heavy metal concentrations in both water and sediment along the course of the Bilina River in the Czech Republic, directly attributed to extreme rainfall events. However, it is important to note that floods had a comparatively shorter-lived influence on the quality of water and sediment compared to the cumulative impacts of anthropogenic activities.

A study by Supriyantini and Endrawati (2015) also reported elevated concentrations within water bodies, sediment, and tissues of the green mussel (Perna viridis) in the coastal rivers of Tanjung Emas Semarang. In this investigation, the elevated Fe content in the estuary stemmed from various sources,

including soil and anthropogenic activities occurring on land. These activities comprise household waste containing Fe, water reservoirs sourced from Fe, industrial waste deposits, and corrosion from Fecontaining pipes transported 70 the estuary by river flow. According to Haeruddin et al. (2020), the highest heavy metal concentrations within sediment were proximal to contamination sources.

Fe contamination within estuarine waters emanates from natural processes, such as geological weathering, rock decomposition, and anthropogenic activities. These anthropogenic sources include Fe release from ore materials and volcanic eruptions that discharge Fe-rich materials into water bodies through runoff, erosion, and flooding. Human activities such as coal transportation and mining operations contribute significantly to this pollution (Dalu et al., 2020). Anthropogenic Fe sources also include land-based activities, such as the discharge of household waste containing Fe, Fe-laden water reservoirs, industrial waste deposits, and the corrosion of Fe-containing pipes, releasing Fe²⁺ and sulfuric acid due to the oxidation of Fe pyrite (FeS2) present in coal layers (Jaishankar et al., 2014).

The escalating temporal increase in Fe concentrations in both water bodies and sediments, as observed in this study, was supported by an investigation conducted by Weber et al. (2020) concerning the contamination in Brazil's Piranga and Doce rivers. Both rivers were contaminated with Fe due to the collapse of the Samarco mine tailings dam in Brazil in 2015. Environmental catastrophes such as dam failures that release tailings profoundly impacted the landscape and aquatic environments, progressively

deteriorating the water quality of the Doce and Piranga rivers over time. Another source of this contamination stems from mineral weathering/erosion and elevated Fe content in soil. The contamination of these rivers has led to the bioaccumulation of Fe in Hoplias intermedius and Hypostomus affini fish. This caused histopathological lesions in the liver, including vacuolization, hepatocyte hypertrophy, and elevated necrotic areas compared to fish from non-tailingscontaminated rivers. This study strengthens the notion that fish living in heavy metal-contaminated waters experience declining aquatic health quality, leading to heavy metal bioaccumulation and histopathological lesions.

Accumulation of iron in sediments: Multifaceted sources and ecological implications

In this study, sediments in the Kuala Lupak estuary accumulate higher levels of heavy metals than the water column. This has rendered the sediments a significant potential source and a primary repository for all pollutants (Weber et al., 2013). These particles can store diverse heavy metals in substantial and consistent quantities, hence, they become valuable indicators for assessing water pollution status (Haeruddin et al., 2020). Liline et al. (2022) reported that sediments in the Rung, Waai, and Poka mangrove areas in Maluku function as a medium for transferring heavy metals su62 as chromium (Cr) and Fe to mudskipper fish. The heavy metals in estuarine sediments result from sedimentation processes that retain significant amounts of adsorbed metal particles, leading to their deposition at the bottom. Wastes containing these metallic elements were deposited in sediments upon entering estuaries. This was facilitated by reduced environmental conditions, low dissolved oxygen, and ele 58 ed pH, causing metal concentration or eightherent (Carvalho Neta et al., 2019). The result was 7 line with the study by Dalu et al. (2020), stating that heavy metal content in sediment was significantly higher than in the water column, as suspended particles were absorbing it. High Fe content in sediment can 45d to contamination of water bodies, which can have negative impacts on the health of aquatic biota and the overall ecological balance of aquatic ecosystems.

Sediments serve as indicators of land-based pollution by acting as "sinks" for pollutants. The rise in heavy metal levels in sediments suggests that heavy metals in the water column are deposited along with suspended solids, posing a potential threat to estuarine and coastal ecosystems. These sediments, serving as sinks for various human activities, may also act as sources of heavy metals, underscoring the ongoing necessity for monitoring to evaluate their role as pollution sources. Heavy metals are prone to accumulating in sediments, resulting in consistently higher concentrations than those in the water column. Suspended sediment particles quickly transport these metals into water through the movement of water (Weber et al., 2013). The high volume of masses

merchant and coal transport vessels contributes to water mass movement in 563 Barito River estuary.

The accumulation of heavy metal in sediments was attributed to a variety of sources, including leaded gasoline emissions chemical manufacturing industries, motorized transportation, underground pipe corrosi 36 coal-based thermal power plants, and urban waste. The results of this study were in line with the report of Jayaprakash et al. (2015), indicating that heavy metal concentrations in water were consistently lower than in sediment within contaminated water bodies. This is because sediment particles tend to absorb metals from the water. In sediment-rich environments, organisms such as the giant mudskipper fish accumulated heavy metals from their food and water through gills, mouth, and skin mucus, further underscoring the potential for metal bioaccumulation. The intricate interplay between suspended sediment particles and their propensity to absorb heavy metals from the aqueous phase was a crucial mechanism derlying the contrasting concentrations of these metals in water and sediment. This phenomenon accentuated the role of sediments as repositories for heavy metals, specifically in aquatic environments subject to anthropogenic influences and diverse pollutant sources.

The congruence between the report of Jayaprakash et al. (2015) and this study underscores the broader ecological implications off metal accumulation in sediments. The distinct levels of heavy metals within water and sediment reflect the differential interactions between these metals and particulate matter suspended in water. Consequently, sedimentary environments act as crucial hubs for storing heavy metals, providing a pertinent context for comprehending the potential ecological ramifications of anthropogenic pollution in aquable ecosystems. High Fe content in sediment can alter the physical and chemical properties, such as color, texture, and pH. This could also affect the habitat of benthic organisms and the overall ecological balance of aquatic ecosystems (Viana et al., 2021). These results elaborate on the necessity for robust monitoring practices to assess the extent and impacts of contamination within sediment systems.

Direct observations during this study, as well as interviews with residents, led to the identification of sources of coastal swamp water pollution in Kuala Lupak. These sources include residential waste and the waste products of factories situated along the banks of the Barito River estuary. Factories, such as plywood processing plants perform wood washing activities as well as generate solid waste such as wood fragments, bark, sawdust, and sanding residues. Other sources of contamination were believed to arise from gold mining activities in the Barito River basin, the use of paint in shipyards, agricultural activities including the application of pesticides and fungicides, converting mangrove areas to fish ponds, and the application of explosives for fishing. Fuel companies discharging

waste were also suspected sources of pollution. These companies distributed diesel fuel through water transport routes that traverse the Barito River estuary. Additionally, the high density of commercial and coal transport ships, alongside coal spillage and debris entering the river, contributed to the degraded water quality of the river. According to the National Transportation Safety Committee, approximately 7.101 units of coal barges navigate the Barito River (Mitra et al., 2023).

In addition to anthropogenic activities, Fe is sourced from natural processes comprising regular biogeochemical cycles, volcanic activities, soil 46 sion, continental runoff, shelf inputs, as well as atmospheric deposition from volcanic events wildfires (Soltani et al., 2021). Furthermore, as the most dominant transition metal in the Earth's crust, it undergoes multiple natural cycles (Jaishankar et al., 2014). Anthropogenic pollution of trace metals, including Fe, in estuarine waters has deteriorated water quality and negatively impacted fish health (Inayat et al., 2023).

Bioaccumulation potential and ecological impact of iron in giant mudskipper fish

Many studies have documented that indicator species accumulate contaminants in their tissues at higher levels than the surrounding habitat, such as water and sediment. This characteristic makes them valuable as potential bioindicators within ecosys30n health biomonitoring programs. According to Table 2, the concentrations of Fe in the muscle, kidney, liver, and skin of giant mudskipper fish exceeded those in the surrounding aquatic environment. The elevated concentrations within fish tissues were attributed to the increased levels present in both sediment and water. These results correspond with those of Gawad (2018), who posited that high concentrations of heavy metals in sediment and water allow aquatic organisms to accumulate metals biologically and transport them through the food chain. The ability of giant mudskipper fish to accumulate heavy metals in tissues stems from their position at the pinnacle of the food chain. These organisms reside benthically in mud within the intertidal coastal zone and mangrove forest floor; hence, they are susceptible to direct exposure to pollutants within the estuarine environment. According to Ferreira et al. (2019), estuaries represent habitats with continuous and direct exposure to heavy metal pollutants.

Estuaries have gained wide recognition as net sinks for trace metals, including Fe, which precipitate rapidly within the environment (Crerar et al., 1982). While rivers transport substantial amounts of dissolved and particulate Fe into the sea, a significant portion is retained and accumulated 51 hin sediments (Daneshvar et al., 2015; Zhu et al., 2018). This phenomenon is attri 47 d to the exceptional capacity of the sediments as a final trap and adsorptive reservoir for heavy metals in aquatic environments. Fe colloids

are immobilized within surface estuarine sediments and can become mobilized by organic matter and reduction agents. When these values exceed a certain threshold, the mobility and deposition of Fe lead to pollution, thereby posing toxicity risks to fish. This risk arose primarily because of the persistent, nonbiodegradable, accumulative, and diverse sources of trace metals (Shafie et al., 2015). Continued and significant discharges of trace metal pollutants into the sea can damage estuarine and coastal ecosystems, given their chronic toxicity, non-biodegradability, and bioaccumulation in biota, including fish. Trace metals can transfer and biomagnify through food chains, posing severe threats to ecosystem health and human well-being. Furthermore, the contamination of metal within estuarine ecosystems can engender toxic effects in fish and humans, primarily by consuming contaminated food sources (Prasad et al., 2023).

The investigation into the toxic effects of Fe on inhabiting estuarine ecosystems proved a compelling inquiry for biologists in South Kalimantan, attributab to the robust positive correlation between Fe levels in the water, sediment, and fish organs. The heightened concentrations of this metal were suspected to be a consequence of its extensive usage (Bat and Arici, 2018). Furthermore, Fe is a prevalent component of industrial and mining waste, frequently discharged into aquatic environments, hence, exhibiting a pronounced correlation with aquatic biota, including fish. Its elevated contents in water and sediment facilitated a more pronounced accumulation within fish organs. Fe in its ferrous form (Fe²⁺) had heightened toxicity towards fish compared to when it was ferric (Fe3+). Fe2+ exhibited toxic effects by binding to gill surfaces and undergoing oxidation to form insoluble Fe³⁺, thereby inciting cellular damage, ultimately culminating in respiratory dysfunction (Singh et al., 2019).

The bioconcentration of Fe within fish tissues attained its zenith within the liver and gonads, subsequently diminishing in the brain, muscles, and 17 rt (Authman, 2015). The study conducted by Omar et al. (2014) substantiated the liver as a prime target organ for Fe-induced toxicity in fish. Furthermore, the toxicity of this metal within gills disrupts respiration due to the physical obstruction of the surfaces (Dalzell et al., 1999). The alkaline nature of the gill surfaces promoted the facile oxidation of soluble Fe, forming insoluble Fe compounds that cloak gill lamellae, impeding respiratory function.

Excessive Fe absorption led to poisoning in *Labeo rohita*, catalyzing the generation of ROS via Fenton reactions, which damaged biomolecules, cells, and tissues through oxidative stress mechanisms (Singh et al., 2019). This metal 21 potentiates the toxicity of other chemical agents, such as paraquat or 2,3,7,8-55 achlorodibenzo-p-dioxin (Sevcikova et al., 2011). In a study by Kaloyianni et al. (2020), the toxicity of magnetite nanoparticles (Fe₃O₄) was reported in *Danio rerio* and *Carassius gibelio* fish,

evoking oxidative, proteolytic, genotoxic, and apoptotic effects, accompanied by elevated malondialdehyde and carbonyl protein levels.

Conclusion

In conclusion, the study elucidated substantial positive associations in Fe levels across water, sediment, and of the giant mudskipper tissues (Periophthalmodon schlosseri) during the 2020-2022 period. These associations underscored the dynamic interplay between environmental Fe buildup and its assimilation within fish tissues. The graphical representation accentuated a strong correlation (correlation coefficient 1) between waterborne concentrations and those in the liver, muscle, and gills. This indicated that higher Fe levels in water correspond to elevated levels in these fish tissues. Similarly, noteworthy positive correlations were observed between Fe levels in water as well as kidney and skin tissues, with correlation coefficients of 0.97 and 0.98, respectively.

Sediment-bound Fe also correlated positively with kidney, skin, gills, liver, and muscle tissues, with sequential correlation coefficients of 0.90, 0.89, 0.83, 0.82, and 0.81. The strong positive correlation within the various fish organs implied uniform Fe dispersion post-uptake. The skin and gills, primary entry points for this metal, exhibited robust positive correlations with kidney, gills, muscle, liver, and skin tissues. This inquiry into the detrimental impacts of Fe on estuarine fish in South K5 imantan assumed significance due to the strong positive correlations between the concentrations of this metal in water, sediment, and fish organs. The widespread dispersion of Fe from industrial and mining activities underscored its prevalence in aquatic ecosystems, often resulting from industrial and mining by-products infiltrating water bodies.

The intricate associations between Fe 20 The intricate associations concentrations in water, sediment, and fish tissues illuminated the intricate relationship between environmental Fe accumulation and biological assimilation in estuarine ecosystems. The study underscored the importance of comprehending the impact of heavy metal contaminants, such as Fe, on aquatic organisms and habitats, with potential implications for ecosystem health and human wellbeing. Therefore, future studies should comprise a multifaceted approach, including scientific investigations, policy formulation, public awareness, and collaborative endeavors among stakeholders to safeguard the well-being of aquatic ecosystems and human populations.

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