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## Study on Formalization of Oil Palm Trunk Modified by Melamine Formaldehyde

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**Abstract.** Structure, composition and water content of an oil palm wood influence its dynamic-mechanical properties. These are very important to be studied in relation to wood applications. The quality of oil palm wood can be improved through chemical modification, that is by formalization treatment in acid and alkaline conditions followed by an impregnation process using melamine formaldehyde. Variation of formalization time and curing temperature after impregnation on samples were investigated. Based on physical properties and FTIR characterizations, the modification in alkaline conditions is better than that in acidic conditions. The modification in alkaline condition can reduce the water content and increase the density of samples better than those in the acidic condition. The FTIR spectra also evidence this, there are emerging the new functional group and a shift in wave numbers on FTIR spectra with strong absorption peak for the samples after modification.

### Introduction

Oil palm plants that have productive age of 25-30 years must be replaced immediately and enter a period of replanting. Waste from the logging of these oil palm trees are usually burned or stacked and allowed to rot. This problem be a challenge and opportunity for researchers and industries, to convert the waste of oil palm trunks into wooden planks that can be used, among others, as a substitute for conventional wood, furniture, partitions or even musical instruments. However, oil palm trunk is classified as low-quality wood with class levels of wood on IV and V [1] (see Table 1 [2]), high hygroscopic and low dimensional stability [3], hence it is rarely used. This is due to the chemical composition of oil palm trunk, in which the portion of parenchymal tissue is larger than that of the vascular bundle. The vascular bundle composes the outside of the oil palm trunk and supports the mechanical properties of the trunk and also helps in transporting water and nutrients. The middle part of the trunk is composed of 70% of parenchymal tissue that contains hydroxyl groups and many water molecules content in the formation of hydrogen bonds. The vascular amount of the bundle determines the density and mechanical properties of the oil palm trunk [4].

Table 1. Class level of Indonesia wood [2].

Class level of wood	Density (gr/cm <sup>3</sup> )	Modulus of rupture (kg/cm <sup>2</sup> )	Impact strength (kg/cm <sup>2</sup> )
I	> 0.9	> 1100	> 650
II	0.6 - <0.9	725 - <1100	425 - <650
III	0.4 - <0.6	500 - <725	300 - <425
IV	0.3 - <0.4	360 - <500	215 - <300
V	<0.3	< 360	< 215

The quality of oil palm trunk waste can be improved to wood class I, II or III by chemical, heat, and impregnation modification or their combination, by penetrating wood pores with polymeric materials and compacting the wood pores in the vacuum machine which has been set at a certain pressure and temperature [4]. To get a good quality of the logs, the lignin and hemicellulose content must be as minimal as possible, because lignin and hemicellulose are amorphous heteropolymers. This content prevents the wood structure from becoming hard, impermeable and susceptible to microbiological disturbances and oxidative stress [5]. Modification of lignin with other polymers is another way to have changes in the wood structure, that can enlarge its application and improve its product performance. Lignin is a macromolecule compound and more reactive than cellulose or others natural polymer. The reactivity of lignin is determined by its particular structure that can be seen from specific functional groups. Lignin is reacted to formaldehyde and makes a cross-linking reaction [6].

This paper is aimed to study on the relationship between the structural change of the oil palm wood structure due to chemical modification and their physics properties. Chemical modification is divided into two steps, first is the formalization by using formaldehyde solution in acid and alkaline conditions, and the second is the impregnation using melamine formaldehyde resin. Chemical modification of the wood cell-wall polymers has been shown to affect the vibrational properties of wood [7], improve the dimensional stability, reduce the moisture absorption and mechanical strength [8].

### Experimental

**Material preparation.** Wood samples were from Banjarbaru of South Kalimantan over 30 years old. The samples were cut into the dimension (2 x 5 x 15) cm that was selected one meter from the ground. All of the samples were initially dried at 60°C for 24 hours to prevent shrinkage and later dried at 103±2°C for 24 hours to get a moisture content of less than 20%. The dried samples were stored in a room temperature before treatment. The average moisture content of the raw sample is 107.03%.

**Modification materials.** Formaldehyde solution, 37 wt%, NaOH 3 wt% and melamine formaldehyde resin supplied from commercial products.

**Modification process.** The dried samples were modified in two steps, formalization with a formaldehyde solution and then impregnated with a melamine formaldehyde resin. For investigating the effect of the formalization on the oil palm trunk toward to changes of structure, formalization was carried out in two conditions, in acidic (pH 4) and alkaline (pH 10), with the immersion times of 1, 3, 5 and 6-days. Afterwards, the samples were dried in the air for seven days or until the samples did not smell formaldehyde anymore. The next step, for improving the physics properties, the specimens were impregnated with melamine-formaldehyde resin by using immersion in a closed container for 4-days to obtain a maximum absorption [8]. After the impregnation treatment, all specimens were air-dried for 1-day before being cured at a temperature of 100°C and 120°C for 1 hour for polymerization. Finally, the samples were dried in an oven 103 ± 2°C for 24 hours to achieve a water content of <20% for the investigation [9].

Water content is a percentage of the oven-dry mass of wood. Water contents will be higher than 100% if the mass of water exceeds the oven-dry mass of the wood. This occurs in green wood of many lower density species. To determination of water content of samples before and after treatment was calculated by using Eq. 1 by SNI 01-4449-2006 [9].

$$KA = \frac{BA - BKO}{BKO} \times 100\% \quad (1)$$

where,  $KA$  is the water content,  $BA$  is the initial weight and  $BKO$  is the oven dry weight.

The densities of the oil palm trunk samples, before and after the modification treatment, were determined in accordance with SNI01-4449-2006 [10] in Eq. 2.

$$\rho = \frac{m}{V} \quad (2)$$

Where  $\rho$ ,  $m$ , and  $V$  are density ( $\text{gr}/\text{cm}^3$ ), mass and volume of the sample respectively.

The structural change of the oil palm trunk was analyzed by Fourier transform infrared spectroscopy (FT-IR), Bruker Alpha RT-DLaTGS, with plate attenuated total reflection (ATR) to investigate functional group presented in the untreated and treated samples.

## Results and Discussions

The effects of modification treatment for physical properties of the oil palm trunk samples were given in Table 2. The formalization conditions treatment followed by the impregnation process can increase the density of samples. Modification in acidic conditions with curing temperatures of  $100^\circ\text{C}$  can increase the density of untreated sample from  $0.37 \text{ gr}/\text{cm}^3$  to  $0.78 \text{ gr}/\text{cm}^3$  and  $0.63 \text{ gr}/\text{cm}^3$ , but when the curing temperature is raised to  $120^\circ\text{C}$ , the density decreases to  $0.37 \text{ gr}/\text{cm}^3$  and  $0.48 \text{ gr}/\text{cm}^3$ . It was caused by water content contained in the wood, at when the curing temperature increases the water in the samples also evaporates. Contrast to the alkaline treatment, the density of the samples after treatment increased with almost the same value, both in the curing temperature of  $100^\circ\text{C}$  and  $120^\circ\text{C}$ . This is indicating that the polymerization occurs between melamine formaldehyde and the O-H group in the wood structure.

Table 2. Physics properties of the oil palm trunk modified.

Formalization time (day)	Curing temperature ( $^\circ\text{C}$ )	Water content (%)		Density ( $\text{gr}/\text{cm}^3$ )	
Untreated		15		0.37	
Treated		Acid condition	Alkaline condition	Acid condition	Alkaline condition
3	100	15.95	10.62	0.78	0.55
5	100	13.37	9.26	0.63	0.65
3	120	10.6	6.19	0.37	0.52
5	120	8.93	5.82	0.48	0.51

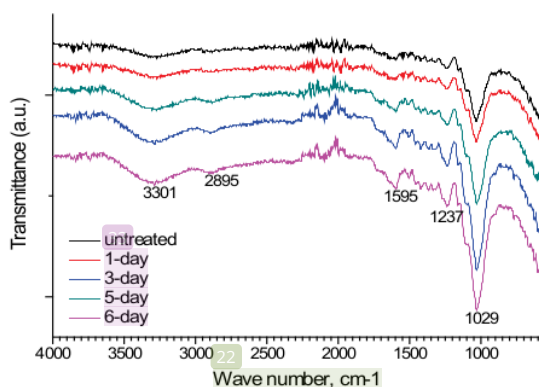


Fig. 1. FTIR spectra of the oil palm trunk samples at different formalization time.

The structural changes of oil palm wood, before and after formalization treatment, shown in Fig. 1. The FTIR spectra of 1-day formalization have the same tendency as the untreated sample, and the 3-day formalization has the same trend as the samples with 6-day formalization. So, for further studies, we used the samples with formalization treatment at 3 and 5-days. After formalization, the peak 2885-2921  $\text{cm}^{-1}$  presence in the formalization 3, 5 and 6-days. This is associated with the functional group of methyl (formaldehyde) C-H stretching on aliphatic lignin. The peak of wave number 3293-3325  $\text{cm}^{-1}$  is related to the vibration of O-H stretching which shows that the sample is hydrophilic. The peak 1595  $\text{cm}^{-1}$  is related to aromatic skeletal vibration and C=O stretching associated with vibrations in lignin. The peak of 1229-1243  $\text{cm}^{-1}$  is the syringyl ring C-O vibration stretching functional group associated with the acetyl group in lignin, and the peak of 1025-1033  $\text{cm}^{-1}$  is related to C-O-C stretching in polysaccharides. The peak intensity of the functional groups is getting negative with increasing formalization time because of the interaction of hydroxyl group of wood with formaldehyde [11].

The FTIR spectra of the oil palm trunk modified by melamine formaldehyde in the acid condition are shown in Fig. 2. The cross-linking in the structure of wood modified by melamine-formaldehyde is more clearly shown in the samples with formalization for 3-days cured at 100°C. Some peaks were not found in other samples. Peaks of 1640  $\text{cm}^{-1}$  and 1535  $\text{cm}^{-1}$  associated with the absorption peak of the triazinyl and the peak of 1375  $\text{cm}^{-1}$  corresponds to the methylene C-H bending vibration for deformation of the polysaccharides [11,12]. The peak 809-815  $\text{cm}^{-1}$  presence in the spectra of 3-day formalization cured at 100°C and 120°C, and the samples with 5-day formalization cured at 120°C. These peaks are attributed to the absorption peak of the triazine ring, which indicates that binding of melamine formaldehyde has occurred to the surface of the cell wall of wood [9,13]. The absorbance peak of this spectra is weak, and this is indicated that penetration melamine formaldehyde to the cell wall of the wood is weak too. The shift of wave number from 1033  $\text{cm}^{-1}$  to 1003  $\text{cm}^{-1}$  occurs in the sample with 3-day formalization cured at 100°C which is caused by the transfer of the hydroxyl group on the wood wall to the polymer chain due to polymerization. It can be seen in Fig. 2. that the intensity of the functional group at 2885-2921  $\text{cm}^{-1}$  is getting weak after treatment for all samples. The reduction of this peak indicates that some formaldehyde disappeared from the wood during the curing process.

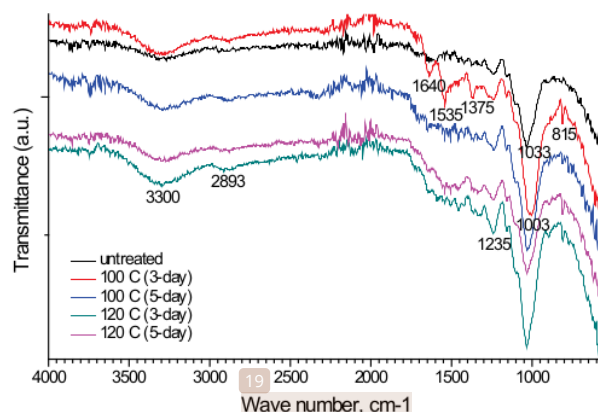


Fig. 2. FTIR spectra of the oil palm trunk samples for the acid condition formalization at the varied curing temperature.

The FTIR spectra of the oil palm trunk modified by melamine formaldehyde in an alkaline condition are shown in Fig. 3. The figure shows that cross-linked of melamine formaldehyde in the cell wall of the wood in alkaline condition is better than that in acid condition. The absorption peak of triazine ring at 807-813  $\text{cm}^{-1}$  presences in all samples and becomes prominent in the samples with the 3-day formalization cured at 100°C and 5-days formalization cured at 120°C. The peaks of



3293-3319  $\text{cm}^{-1}$  are related to the vibration of O-H stretching and stretching vibration of secondary amine [12]. The reduction of the peak intensity for samples with 3-day formalization cured at 120°C indicates that hydroxyl group is significantly diminished after modification because crosslink melamine formaldehyde is weak. For the sample with 3-day formalization cured at 100°C, there is also a shift in a wave number from 1033  $\text{cm}^{-1}$  to 1001  $\text{cm}^{-1}$  caused by polymerization. The presence of strong intensity of the peaks at 1375  $\text{cm}^{-1}$  and 1551  $\text{cm}^{-1}$  in the samples with 3-day formalization cured at 100°C and 5-days cured at 120°C indicates that those samples are the most thoroughly impregnated. This shows that the reaction of melamine formaldehyde to the walls of wood cells is more absorbed in alkaline condition. Finally, for this modification, only the sample with 3-day formalization cured at 100°C which still had an aldehyde functional group after the curing process.

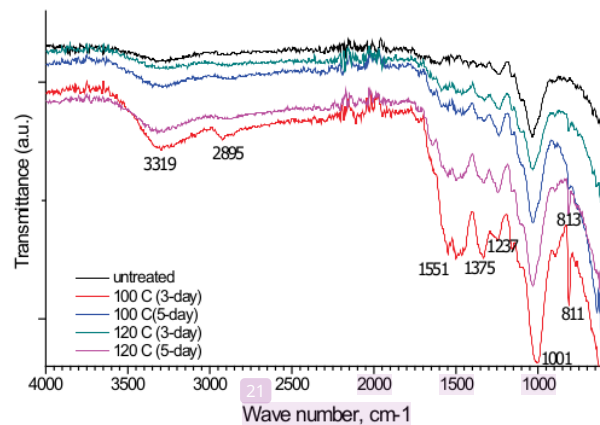


Fig. 3. FTIR spectra of the oil palm trunk samples for the alkaline condition formalization at the varied curing temperature.

## Conclusion

Improving the physical properties of oil palm trunk by changing its wood structure has been carried out by the chemical modification method. The modification in an alkaline condition can reduce the water content of wood better than that in an acidic condition. Increasing the density of oil palm wood is also better in alkaline conditions for all curing temperatures. In acidic conditions, the water content is still high if the samples are cured at 100°C and decreases when curing temperature is raised. This situation occurs because the absorption of melamine-formaldehyde on the cell wall is not impregnated thoroughly, as evidenced by the presence of triazine ring group with weak intensity, and this is in contrast to the alkaline formalization. Modification in an alkaline condition is found to be more effective since formaldehyde compounds can be completely removed after the curing process, except for the samples with the formalization of 3-days with a curing temperature at 100°C. Based on the value of physical properties and the changed in the chemical composition of the samples, we can be concluded that the modification in alkaline conditions is better than that in acid conditions.

## Acknowledgements

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