From: ICoSiET 2020 Sent: 25 July 2020 22:18 To: Yulian Firmana Arifin Subject: ICoSiET 2020 submission 17

Dear authors,

We received your submission to ICoSiET 2020 (2020 International Conference on Science in Engineering and Technology (ICoSiET)):

Authors : Yulian Firmana Arifin, Muhammad Arsyad, Muhammad Afdi and Husaini Muslim Title : The Application of Vapour Equilibrium Technique to study the Shrinkage and Water Retention of Compacted Claystone-Bentonite Mixtures Number : 17

The submission was uploaded by Yulian Firmana Arifin <<u>y.arifin@ulm.ac.id</u>>. You can access it via the ICoSiET 2020 EasyChair Web page

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Thank you for submitting to ICoSiET 2020.

Best regards, EasyChair for ICoSiET 2020. From: ICoSiET 2020
Sent: 25 September 2020 19:32
To: Yulian Firmana Arifin
Subject: ICoSiET 2020 notification for paper 17

Dear Yulian Firmana Arifin, Muhammad Arsyad, Muhammad Afdi, Husaini Muslim

Congratulations! Your paper 17 ('The Application of Vapour Equilibrium Technique to study the Shrinkage and Water Retention of Compacted Claystone-Bentonite Mixtures') has been accepted and will be presented in The 2020 International Conference on Science in Engineering and Technology (ICoSiET). ICoSiET 2020 will be held on October 21-22, 2020 in Palu, Indonesia. However, We are delighted to inform you that we will use Zoom to deliver our online conference if the participant doesnt possible to attend the conference due to COVID-19 Outbreak.

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4. Make sure that you have registered yourself as a speaker through the conference website: http://icosiet.org/2020/registration/

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Important Dates:

Camera ready/Revised Paper Version: September 30, 2020. Early Bird/Payment Registration: September 30, 2020. Regular Regsitration: October 5, 2020.

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Thank you for choosing the ICoSIET 2020 to present your research results. We are looking forward to welcoming you at this conference.

Regards, Technical Program Committee

SUBMISSION: 17

TITLE: The Application of Vapour Equilibrium Technique to study the Shrinkage and Water Retention of Compacted Claystone-Bentonite Mixtures

----- REVIEW 1 -----

SUBMISSION: 17

TITLE: The Application of Vapour Equilibrium Technique to study the Shrinkage and Water Retention of Compacted Claystone-Bentonite Mixtures

AUTHORS: Yulian Firmana Arifin, Muhammad Arsyad, Muhammad Afdi and Husaini Muslim

----- Overall evaluation ------

SCORE: 2 (accept)

----- TEXT:

This paper is well written and structured, however some improvements need to address:

- The word "study" in the title and abstract needs to write more specific. What the Vapour Equilibrium Technique for in this study?

- In the abstract needs to address the implication of benefits to the field of this study.

- The reviewer suggest the structure of the paper should be Introduction, Method, Results and Discussion, and Conclusion.

- Rewrite the conclusion into one paragraph.

<u>The Application of Vapour Equilibrium Technique to study the Shrinkage and Water Retention of</u> Compacted Claystone-Bentonite Mixtures

Yulian Firmana Arifin, Muhammad Arsyad, Muhammad Afdi, Husaini Muslim

Abstract

The <u>vapour</u> equilibrium technique is widely used to study the behaviour of soil experiencing dryingwetting phenomena. It is considered to be inexpensive, simple, and has the ability to adequately control the suction applied to soil samples. This <u>paper</u>, therefore, describes its application in studying the shrinkage and water retention in compacted claystone-bentonite mixtures. This involved using five saturated salt solutions including potassium sulphate (K₂SO₄), potassium chloride (KCl), sodium chloride (NaCl), potassium carbonate (K₂CO₃), and magnesium chloride (MgCl₂.6H₂O). The sample was allowed to be in equilibrium with the relative humidity salt <u>solution and a calliper</u> was used to measure the dimensions every day up to when this was achieved. The results showed the bentonite in the mixture j affects the amount of shrinkage and water retention while the sample's initial moisture content was also found to be very influential on the magnitude of the primary and residual shrinkage. Moreover, the sample's ability to hold water was almost the same without differentiating the initial water content <u>at a</u> total suction of more than 39055.36 kPa.

Keywords: shrinkage, water retention, vapour equilibrium technique, claystone-bentonite mixture

Introduction

Claystone is a material <u>usually</u> avoided and discarded <u>due to</u> its unfavourable <u>behaviour</u> and the existence of its layers considered to be a problem in construction <u>work have been</u> reported in some publications [1– 5]. The use of this material has, however, been widely recommended, especially as a barrier to hazardous <u>waste and this requires mixing it</u> with bentonite to improve its hydro-mechanical properties [6–9]. Zang and Kröhn [9] reported the superiority of the claystone-bentonite mixture's hydro-mechanical properties compared to compacted pure bentonite and bentonite-sand <u>mixture</u> which included higher density, stiffness, and stability, and very low <u>permeability as well as its</u> ability to release gas at <u>lower</u> pressure [9]. The characteristic absorption of clay minerals in <u>the</u> claystone/clay shale [10] <u>also</u> adds to its advantage compared to the use of sand to mix the clay liners. It is even more beneficial when the claystone is obtained at the project site <u>due to the consideration for</u> the use of local material by several researchers in recent times [11–13].

<u>The two mixtures are made from clay and one of the characteristics considered is their high shrinkage</u> which was divided by Cornelis et al. [14] into four parts which are structural, normal, residual, and zero. Meanwhile, Mishra et al. [15] <u>also</u> classified shrinkage in the compacted clay and bentonite-sand mixtures into initial, primary, and residual stages. <u>It is important to note that both structural and initial shrinkages</u> are influenced by the bimodal pore distribution in the form of macro and micropores which were reported / by Arifin [16] to be strongly influenced by the initial moisture content of the samples. Generally, the / initial and residual shrinkages are not too significant and this means there is a need to study the primary shrinkage in claystone and its mixture with bentonite.

Mishra et al. [15] <u>conducted used a drying process to obtain a shrinkage curve and this involved</u> allowing the sample to be <u>air-dried for two months. It is, however, difficult to implement this method in areas with</u> high relative humidity (RH) such as in <u>Indonesia due to the possible absorption of water from the air by</u> the <u>sample when the RH rises</u>. <u>Therefore, a</u> better method is to control the RH around the sample using the Vapor Equilibrium Technique (VET) to ensure the sample is <u>balanced</u> at a lower <u>RH</u>. <u>This approach</u> Deleted: keywords

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has also been widely applied to test water retention behaviour or water characteristic curves of soils, especially clay apart from its application in shrinkage behaviour [16–18]. It has also been applied in testing the hydro-mechanical behaviour of unsaturated compacted clay to control the suction applied to the sample [7,16,19–21]. VET, however, generally requires using a saturated salt solution with some researchers observed to have used molal salt solutions such as sodium chloride [16,18]. The sample also needs to be in equilibrium with the relative humidity above the solution [7,16–20,22] and this means temperature needs to be controlled due to its strong influence on the RH. The method is also usually combined with the axis translation technique at low suction [16,21] and the changes in the temperature gradient affect the RH with further causes a variation in the total suction being applied [16,23,24]. This paper, therefore, discusses the shrinkage behaviour and water retention of compacted claystone-bentonite mixtures using VET.

Materials used

The materials used in this study include claystone and bentonite with the claystone obtained from the Banjarbaru city area while the bentonite is available in the market. The Engineering Properties of the two materials are summarized in Table 1 and the claystone observed to have a liquid limit (LL) of 50.76% and a plasticity index (PI) of 20.95% while and these are twice as much as those obtained from Belencito formation in eastern Andes mountain, Columbia, as reported by Espitia et al. [1].

Properties		Claystone	Bentonite
Specific gravity		2.60	2.71
Water content	%	2.75	14.17
Grain size distribution:			
Gravel (>2 mm)	%	0.02	0.00
Coarse sand (0.6-2.0 mm)	%	0.08	0.00
Medium sand (0.2-0.6 mm)	%	0.10	0.00
Fine sand (0.05-0.2 mm)	%	4.30	1.39
Silt and Clay (0.002-0.05)	%	43.94	8.33
Clay (<0.002mm)	%	51.55	90.28
Atterberg Limits:			
Liquid limit	%	50.76	351.71
Plastic limit	%	20.95	44.68
Shrinkage limit	%	9.74	41.89
Plasticity Index	%	29.81	307.03

 Table 1. Engineering properties of materials used

Methods and Procedures

Samples Preparation

The Standard Proctor compaction test (ASTM D698-07) was <u>conducted to determine the relationship</u> between water content and claystone density <u>used as a primary material and the optimum moisture</u> content and maximum density were <u>found to be 15%</u> and 16kN/cm³, respectively. <u>All the samples were</u>

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study the effect of water	content.	er and nigher w	ater contents at 1	0% and 15% were us	ed to	Deleted: than optimum water contents (i.e.,t 10% at [5]
Claystone was mixed wit to reach the pre-selected sample was also calcula statically compacted to r usual as observed with th	h 5%, 10%, 15%, and Jevel of 10%, 15%, a ted before mixing. The each a density of 16 e 10 mm height and 6	1 20% <u>bentonite</u> nd 20% and valu he <u>mixed</u> sampl kN/ <u>m³ and the</u> 4.2 <u>mm diamete</u>	on, dry, weight ba 10. Meanwhile, th 10. Weanwhile, th 10. Weanwhile, th 10. Weight back 10. Weight bac	sis while water was a ne water contained in allibrium for one day ras, however, thinner	dded each and than	Deleted: and bentonite are was mixed with bentonite c [6]
There are two possible or the <u>experience of an incr</u> wetting. <u>It is, however, r</u> <u>leading to shrinkage. The</u> <u>vapour equilibrium techn</u>	onditions in the clay_l ease in water content ossible to also experi sample was simulate ique (VET) method.	iner or cover la due to rain in th ence immediate d in this study a	yer while on the f e sample which n drying due to the nd this led to the	field and the first invo neans the initial proce e exposure to air, the second condition wit	blves ess is ceby, h the	Deleted: that occur in the clay liner layerr cover lay[7]
The total suction of the sensor corrected for accube the most reliable tool using Equation 1 while the 2.	initial sample was det racy using the chilled to measure RH [16,2 he initial conditions of	ermined <u>before</u> -mirror hygrome 3–25]. <u>Meanwh</u> <u>5 the</u> claystone a	being tested by n eter technique wh ile, the total suction ad bentonite mixt	neasuring the RH usi ich has been confirm ion sample was calcu ture are presented in T	ing a ed to lated 'able	Deleted: Before being tested, the The total suction of t [[8]
$s_t = -\frac{\rho_{W.RT}}{M_w} \ln\left(\frac{RH\%}{100}\right)$				(1)		
where s is total suction is 8.31432 J/mol.K, T which is 18.016 g/mol.	ρ_w is the volumetri is the temperature	c weight of wa in Kelvin, and	ter, $\frac{\mathbf{R} \text{ is a }}{M_w}$ is molecule	ersal gas constant we mass of water va	hich pour	Deleted: whereas where s is total suction, $\rho_{w} = \dots \rho_{w}$ is $\left(\dots [9] \right)$
Tal	le ? Initial conditio	ns of clayston	hentonite mixt	lirec		- Deleted: condition
Sample code	Bentonite content	Dry density	Water content	Total suction	^ '	
10005 1	(%)	(kN/m^3)	(%)	(kPa)		
100CS-1	0	16	10	2005.50		
100CS-3	0	16	20	1816 35		
95CS-5B-1	5	16	10	4089.92		
95CS-5B-2	5	16	15	2664.36		
95CS-5B-3	5	16	20	2104.87		
90CS-10B-1	10	16	10	5231.79		
90CS-10B-2	10	16	15	4233.12		
90CS-10B-3	10	16	20	3236.72		
Vapour <u>Equilibrium Tec</u>	hnique					Deleted: equilibrium technique
<u>The five</u> saturated salt so (KCl), sodium chloride (<u>The RH</u> or total <u>suction</u> c	lutions used in this stu NaCl), potassium car of each solution calcul	ady include pota conate (K2CO3) ated <u>using</u> Equa	ssium sulfate (K2), <u>and magnesium</u> tion 1 <u>js s</u> ummari	2SO4), potassium chlo 1 chloride (MgCl2.6H zed in Table 3.	oride 20).	Deleted: Five The five saturated salt solutions were . [10]
Т	able 3. <i>RH</i> and total	suction of satu	rated salt soluti	on		
Solution	K_2SO_4	KCl Na	$Cl K_2CO_3$	MgCl _{2.} 6H ₂ O		

RH (%)	95.2	81.5	74.3	46	34.2
$T(^{\circ}C)$	27.7	28.1	27.8	28.2	28.0
Total suction (kPa)	6801.21	28318.56	41084.91	107528.77	148484.33

The solution <u>was later</u> placed in an airtight glass equipped with rubber seals and clamp <u>lids</u> as shown in Figure 1. <u>The VET was generally conducted</u> using a desiccator [17,22] and, apart from the larger volume, the samples placed <u>in the desiccator influenced</u> each other and <u>this slowed</u> down the time to reach equilibrium [16].



the height and diameter of the samples were measured every day using Vernier callipers through the direct measurement method [26]. They were placed in the oven for 24 hours to obtain the dry weight and two samples were tested to represent the conditions presented in Table 1. Moreover, smaller glasses were used to ensure each is filled with different solutions and due to its ability to allow concurrent testing of identical samples to reduce the time needed for the test.

Results and Discussion

Time to reach equilibrium

The samples were considered to have reached equilibrium condition when no change was recorded in their weight and dimensions. Figure 2 shows a typical change in the weight, diameter, and height of the samples 95CS-5B in (a) and 90CS-10B in (b) for a glass containing a saturated K2SO4 solution with a total suction of 6848.64 kPa, This means the weight decreases by time followed by the diameter and height. Meanwhile, the weight at 95CS-5B-3 and 90CS-10B-3 with an initial water content of 20% was constant on the 11th and 12th day while the value was also observed to be constant on the 2th day for 95CS-5B-1 and 90CS-10B samples -1 were relatively constant at the 9th and 10th days while it was 11th and 12th day of 95CS-5B-3 and 90CS-10B-3. These results showed the equilibrium time was affected by the initial water content such that a longer time is required to achieve equilibrium at a higher value for the initial moisture content.

Figure 3 shows the change in the weight by time at higher total suction of 39055.36 kPa for 95CS-5B in (a) and 90CS-10B in (b). The weights of the 95CS-5B-1, 95CS-5B-2, and 95CS-5B-3 were observed to



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Shrinkage behaviour of compacted claystone-bentonite mixtures

The soil shrinkage behaviour is usually depicted in the relationship between void ratio and water content	Deleted: Usually, The soil shrinkage behavior cha [19]
[14-16_26], and this is shown for claystone and bentonite mixture in Figure 4. The data is consistently	
displayed using different shape markers and colours as observed in the triangles, squares, and circles	
applied for the initial moisture contents at 10%, 15%, and 20% respectively. Moreover, green, blue and	
red <u>colours were used to distinguish</u> the percentage of bentonite at 100% claystone, 5% bentonite, and /	
10% bentonite respectively. The sample shrinkage curve is shown in the figure to only have two distinct /	
stages which are primary and residual with no initial shrinkage obtained for all the samples tested. /	
Generally, a sample with three stages usually has a high initial moisture content or become saturated in	
the swelling process as reported by Mirsha et al. [15]. A two-stage shrinkage sample was, however, found	
for the bentonite-sand mixture at optimum moisture content and maximum density conditions	
The water content at which the primary shrinkage ended or the residual shrinkage started (w, as well as	Deleted: end ended or the residual shrinkage begins [[20]]
the magnitude of <u>each</u> , are summarized in Table 4. Moreover, the w _r , PS, and RS are used to represent the	
water content at residual shrinkage, primary shrinkage, and residual shrinkage, respectively. The	
information on Table <u>4 shows</u> samples 100CS-1, 2, and 3 with an initial water content of 10%, 15%, and	
20% have a w_r of 2,068%, 4,196%, and 4.47% respectively. The content at residual shrinkage was /	
observed to have increased with the initial value and this was also the same for the claystone samples with /	
bentonite such as 95CS-5B and 90CS-10B	
Table <u>4 also indicates</u> the influence of the initial moisture content through the PS and RS. The 100CS-1,	Deleted: In Table 4 also indicates the influence of [211]
2, and 3 experienced primary shrinkage of 19.79%, 21.37%, and 35.65%, and residual shrinkage of 0.9%,	
4.74%, and 7.99%, respectively. At the same density, the PS and RS were recorded to have increased with	
the initial water content and the same trend was also obtained for claystone-bentonite mixtures.	
A comparison was made on the samples to determine the effect of bentonite at the same initial moisture	Deleted: To see the effect of bentonite, a A comparis [22]
content. The wr of 100CS-1, 95CS-5B-1, and 90CS-10B-1 at initial content of 10% were 2,068%,	
2,827%, and 6 <u>16 while</u> the primary shrinkage was 19.79%, 16.79%, and 12.40% and the residual	
shrinkage was 0.9%, 2.19%, and 3.39% respectively. These data showed the wr and PS increased while	///
RS_decreased with an increase in the bentonite percentage. The samples with 0% and 20% content also	//
showed the change in void ratio at the same initial water content as shown in Figures 4(a) and (b).	/

Table 4. Shrinkage properties of claystone-bent	onite mixtures

Table 4.	Shinkage	properties	of claystol	ie-bentonn	e mixtures				
		100CS			95CS-5B			90CS-10B	
	1	2	3	1	2	3	1	2	3
w_r (%)	2.068	4.196	4.478	2.827	4.313	4.542	6.160	6.436	6.725



The water retention curves of claystone and bentonite <u>mixtures</u> are <u>indicated</u> in the relationship between water content and total <u>suction</u> as shown in Figure 5. In consistence with the previous figures, the 100CS, 95CS-5B, and 90CS-10B are displayed in green, blue, and red respectively while the initial condition is in black. Moreover, <u>those</u> with an initial moisture content of 10%, 15%, and 20% are presented in triangle, square, and <u>circle respectively</u>. Figure 5 (a), <u>therefore</u>, shows the water content of J00CS-1, 2, and 3 was initially 20%, 15%, and 10% and later decreased as the suction was increasing. Meanwhile, the amount of water content retained by the sample is almost the same at the total suction of 41084.91 kPa and the same <u>trend is observed</u> in other <u>samples</u> as shown in Figures 5(b) and (c). Arifin [16] found a similar result for bentonite and bentonite-sand mixtures. <u>Furthermore</u>, dry samples from saturated <u>conditions</u> have almost the same water content at suction higher than the air entry value (AEV) even though <u>they are compacted</u> at different water levels.

The difference in water retention curve at low total suction is <u>due to</u> the pore size distribution from the samples compacted at lower and higher moisture content than the required optimum. At the lower value, / the macropores became dominant while it was the micropores at high values [16]. Moreover, water does / not only fill the intra-particles or micropore but also the inter-particles or macropores at low total suction / and also affected by the fabric. However, at the high total suction of 39055.36 kPa recorded in this study, the water was in the intraparticle or micropore and this reduced the influence of the fabric due to the adsorption of the water with hydration force by the clay surface.





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Samples with the same conditions are plotted in a graph as shown in Figure 6 to determine the effect of the bentonite with those at 10% presented in Figure 6 (a) and 20% in (b). The curve of the sample containing 5% bentonite is observed to be above the curve for the sample without bentonite or 100CS while the highest curve was recorded with 10% content. Moreover, the water content was found to have increased with the bentonite content at the same total suction and similar behaviour is present in Figure 6 (b) showing samples with higher bentonite content, have the ability to absorb more water.



Figure 6. Bentonite effect on water retention behaviour of samples with <u>the</u> initial water content of (a) 10% and (b) 20%.

Conclusions

The conclusions drawn from the shrinkage properties and water retention of compacted claystone and bentonite mixture are as follow:

- 1. The sample shrinkage curve has only two distinct stages which are the primary and residual shrinkage.
- 2. Water content at residual shrinkage was observed to have increased with the initial water content.
- 3. The amount of primary and residual shrinkages also increased with the sample's initial moisture content at the same density.
- 4. <u>The Primary shrinkage also increased with the bentonite content while the residual shrinkage was observed to have decreased</u>.
- 5. The water retention curve was strongly influenced by the initial moisture content at a total suction lower than 39055.36 kPa while the ability to hold water was almost the same at higher values.
- 6. The addition of bentonite <u>to the claystone material</u> increases the ability of the sample to hold <u>water and</u> <u>a higher bentonite content holds more water content at the same total suction</u>.

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Acknowledgment

The authors are grateful for the financial support from the University of Lambung Mangkurat through "Dosen Wajib Meneliti" Program.

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From: ICoSiET 2020Sent: 14 October 2020 6:23To: Yulian Firmana ArifinSubject: [ICoSiET 2020] Acceptance Notice of the Paper 17

Dear Yulian Firmana Arifin, Muhammad Arsyad, Muhammad Afdi, Husaini Muslim

Congratulations! Your paper 17 ('The Application of Vapour Equilibrium Technique to study the Shrinkage and Water Retention of Compacted Claystone-Bentonite Mixtures') has been fully accepted and will be presented in The 2020 International Conference on Science in Engineering and Technology (ICoSiET). ICoSiET 2020 will be held on October 21-22, 2020 in Palu, Indonesia. However, We are delighted to inform you that we will use Zoom to deliver our online conference if the participant doesn't possible to attend the conference due to COVID-19 Outbreak.

Make sure that you have registered yourself as a speaker through the conference website: http://icosiet.org/2020/registration.

The other queries that you have to do are: 1) Prepare your presentation file for 15 to 20 minutes duration (in *.PDF, *.PPTX, *.PPT, or URL Video).

2) Prepare your FINAL PAPER (in an editable file, such as in *.DOC, *.DOCX format). Ensure the similarity score is less than 20%, the images/tables in good resolutions, and you have proofread the English quality carefully.

Both files should be uploaded to https://forms.gle/kTZrN9RC7f5hD2CWA before October 18, 2020.

The schedule of the program and the technical information of the conference will be updated soon and send to your email.

Thank you for choosing the ICoSIET 2020 to present your research results. We are looking forward to welcoming you to ICoSIET 2020.

Best Regards,

Dr. Eng. Rifai General Chair of ICoSiET 2020 From: info@icosiet.orgSent: 26 October 2020 0:46To: y.arifin@ulm.ac.idSubject: [ICoSIET 2020] Certificate of Participation

Dear Yulian Firmana Arifin

On behalf of the ICoSIET 2020 conference organizers, we would like to express my gratitude for your participation to submit and attend our conference virtually which is hosted by Universitas Tadulako, Indonesia.

The event might not be successful without your participation.

As a form of appreciation, we have provided your certificate as presenter enclosed in this email.

Again, we would like to thank you and we are really looking forward to your participation in our forthcoming 2021 2nd ICoSIET conference.

Kind regards,

Dr. Eng. Rifai Mardin General Chair of ICoSIET 2020 From: info@icosiet.org
Sent: 08 October 2021 3:27
To: info@icosiet.org
Subject: [ICoSIET 2020] Publication Progress of ICoSIET on IOP Publishing

Dear Authors,

Thank you very much for your participation in The 2020 International Conference on Science in Engineering and Technology (ICoSiET) which will be held Virtually on October 21-22, 2020.

We deeply apologize for the delayed publication progress of the ICoSIET 2020 proceedings on the IOP publishing scheduled on the 3rd quarter of 2021 (<u>https://ioppublishing.org/mse-forthcoming-volumes/</u>) due to some reasons. However, currently, we are on finalizing the proceedings and expected will be published soon within 1-2 months.

We will update soon if there is any progress of this publication. Thank you very much for your understanding.

Hope you keep safe and healthy during this time.

Best Regards,

Technical Program Committee